

## LEAF FLAVONOIDS OF *ZIZIPHUS SPINA-CHRISTI*

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**Key Word Index**—*Ziziphus spina-christi*; Rhamnaceae; flavonol glycosides; quercetin 3-xylosyl(1 → 2)rhamnoside-4'-rhamnoside.

**Abstract**—From the leaves of *Ziziphus spina-christi*, the new flavonoid quercetin 3-xylosyl(1 → 2)rhamnoside-4'-rhamnoside as well as rutin, hyperin and quercitrin were characterized. The structures were established by chromatography, chemical degradation and UV spectroscopy, and confirmed by  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectroscopy.

### INTRODUCTION

*Ziziphus* species are used in traditional medicine, especially for treating insomnia [1]. Chemical studies have shown the presence of cyclopeptide alkaloids in the bark of *Z. sativa* [2], of saponins in the stem of *Z. mauritania* [3] and of C-glucosylflavones in the seeds of *Z. vulgaris* [1]. In a previous analysis *Z. spina-christi* was shown to contain betulinic and ceanothic acid [4], but there are no reports on other constituents.

The present paper describes the isolation of the new natural product quercetin 3-O- $\beta$ -D-xylopyranosyl(1 → 2)- $\alpha$ -L-rhamnopyranoside-4'-O- $\alpha$ -L-rhamnopyranoside in addition to rutin, hyperin and quercitrin from the aqueous ethanolic extract of the leaves of *Z. spina-christi* Miller. Similar flavonol triglycosides are known to occur in *Rhamnus petiolaris* [5] (Rhamnaceae).

### RESULTS AND DISCUSSION

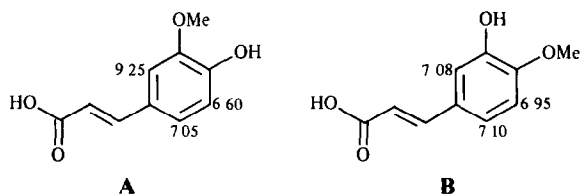
Two-dimensional paper chromatography (PC) of the aqueous ethanolic extract of the fresh leaves of *Z. spina-christi* revealed an oligoglycosylated flavonoid (high mobility in aqueous solvents; dull brown gradually changing to yellow in UV +  $\text{NH}_3$ ), together with a complicated mixture of flavonol 3-glycosides (dark brown changing to yellow in UV +  $\text{NH}_3$ ). Application of polyamide column chromatography (CC) followed by PCC afforded four pure compounds, 1–4.

Acid hydrolysis of compound 1 gave quercetin, rhamnose and xylose. UV spectral analysis of 1 in methanol and in the presence of diagnostic reagents [6, 7] suggested that the sugar moieties are bonded to the quercetin moiety at positions 3 and 4' (positive shift with NaOAc, small shift with  $\text{AlCl}_3$ , small and moderately intense shift with NaOMe, and no shift with NaOAc- $\text{H}_3\text{PO}_3$ ).

Enzymatic hydrolysis of 1 with  $\alpha$ -rhamnosidase [8] gave an intermediate, 1a (dark brown on PC changing to yellow in UV +  $\text{NH}_3$ ), which gave UV spectral data similar to those reported for quercetin 3-glycosides [9]. 1a

yielded quercetin, rhamnose and xylose (coPC) on acid hydrolysis; on controlled acid hydrolysis, it yielded quercetin 3-rhamnoside (quercitrin) (coPC). 1a is therefore quercetin 3-xylosylrhamnoside and 1 is quercetin 3-xylosylrhamnoside-4'-rhamnoside. The suggested structure of 1 was confirmed by NMR spectroscopy. From the  $^{13}\text{C}$  NMR spectra of 1, the presence of two rhamnose moieties followed from two signals in the methyl region. The positions of these signals at  $\delta$ 17.84 and 18.08 indicated that the sugars were attached directly to the quercetin hydroxyls, because attachment to sugar hydroxyls would shift the signals downfield to ca  $\delta$ 21 [10]. The sugar moieties (rhamnose or xylose) must be attached to positions 3 and 4' of quercetin, because these carbon signals were shifted upfield and the corresponding *ortho* and *para*-carbon signals were shifted downfield (see Experimental). Similar shifts are well-known from the work of Markham *et al.* [10]. The  $\beta$ -configuration of the xylose moiety was derived from the C-1 chemical shift at  $\delta$ 106.9 [11]; the  $\alpha$ -configuration of the two rhamnose moieties followed from the C-1 chemical shift values at  $\delta$ 101.0 and 99.2 [10]. Attachment of the xylose moiety to C-2 of rhamnose was indicated by the shift of the rhamnose C-1 signal to  $\delta$ 99.2 ( $\gamma$ -upfield shift caused by C-1 of xylose) and of the rhamnose C-2 signal to  $\delta$ 80.4 ( $\beta$ -downfield shift caused by C-1 of xylose). The chemical shift values of all the sugar carbons confirmed the pyranose form of the three sugar moieties [12].

The  $^1\text{H}$  NMR spectrum of 1 was also in accordance with the proposed structure. The small chemical shift difference ( $\Delta\delta = 0.23$  ppm) between the signals of the C-5' and C-6' protons indicated substitution (methylation,



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acetylation, glycosylation, etc.) on the hydroxyl group at C-4' and not at C-3'. A reversed substitution would have caused a larger shift difference, as can be derived from the  $^1\text{H}$  NMR data (obtained in the same solvent, DMSO) of ferulic acid A and isoferulic acid B.

The two rhamnose anomeric protons gave rise to two signals, the positions of which ( $\delta$  5.34 and 5.37) indicated the attachment of the anomeric carbons of each rhamnose to the quercetin hydroxyls [6], and the half-width of which (*ca* 4 Hz) proved the  $\alpha$ -configuration at the anomeric carbons. The anomeric proton of the xylose moiety was hidden by a broad hydroxyl signal at  $\delta$  4–4.5.

The conformation of the three sugar moieties is  $^1\text{C}_4$  for the two rhamnose moieties and  $^4\text{C}_1$  for the xylose moiety. This follows from the  $\alpha$ - and  $\beta$ -configurations discussed above. The attachment of the disaccharide moiety to C-3 and not to C-4' could not be derived from  $^1\text{H}$  NMR or from  $^{13}\text{C}$  NMR data, but only from the results of enzymatic degradation as discussed above.

Compound 2 was identified as rutin by acid hydrolysis, UV spectral analysis and coPC. The structure was confirmed by  $^1\text{H}$  NMR spectroscopy, which gave data identical to those reported for rutin [6]. Further confirmation of the structure was achieved through  $^{13}\text{C}$  NMR spectroscopy.

Compounds 3 and 4 were identified as hyperin and quercitrin, respectively, by acid hydrolysis, UV spectral analysis, coPC,  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectroscopy [6, 10].

#### EXPERIMENTAL

NMR: JEOL FX 100,  $\delta$  values, solvent DMSO- $d_6$ ; reference: signal of DMSO- $d_6$  set at  $\delta$  39.5 which is the chemical shift in relation to  $\delta_{\text{TMS}} = 0$ . Atomic absorption: Varian 1000 spectrometer. PC was carried out on Whatman No. 1 paper using solvent systems: (1) HOAc (HOAc-H<sub>2</sub>O, 3:17); (2) BAW (*n*-BuOH-HOAc-H<sub>2</sub>O, 4:1:5, top layer); (3) iPW (*i*-PrOH-H<sub>2</sub>O, 11:39); (4) forestal (conc. HCl-H<sub>2</sub>O-HOAc, 3:10:30); (5) BPOH (C<sub>6</sub>H<sub>6</sub>-*n*-BuOH-pyridine-H<sub>2</sub>O, 1:5:3:3, top layer). In addition, solvent systems 1, 2 and 3 were used in PPC on Whatman No. 3 paper.

Leaves were extracted with EtOH-H<sub>2</sub>O, 1:3. The dried extract was transferred to a Polyamide column and eluted with H<sub>2</sub>O followed by H<sub>2</sub>O-MeOH (9:1, 3:2 and 7:3), successively. Pure 1 was isolated from the 9:1 fraction by PPC using solvent 3. Pure 2 and 3 were isolated from the 3:2 fraction by PPC using solvents 1 and 2. Crystals of 4 were separated from the concentrate of the 7:3 fraction.

*Quercetin 3-xylosyl(1 → 2)rhamnoside-4'-rhamnoside* (1).  $R_f$  values: 0.58 (HOAc); 0.50 (BAW); 0.61 (iBW). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 256, 266', 353;  $\Delta\lambda$  (nm) on addition of: NaOAc = 6, AlCl<sub>3</sub> = 52, AlCl<sub>3</sub>-HCl = 50, NaOMe = 45. Acid hydrolysis (1.5 M aq. HCl, 100°, 45 min) of 1 gave xylose, rhamnose (coPC), and quercetin (mp, mmp, coPC and UV data). Hydrolysis with  $\alpha$ -rhamnosidase (pectinase, from Koch and Light) yielded the intermediate

quercetin 3-xylosyl(1 → 2)rhamnoside (1a).  $R_f$  values: 0.52 (HOAc); 0.60 (BAW); 0.55 (iPW). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 256, 266', 295', 352;  $\Delta\lambda$  (nm) on addition of: NaOAc = 6, NaOAc-H<sub>3</sub>PO<sub>3</sub> = 15, AlCl<sub>3</sub> = 68, AlCl<sub>3</sub>/HCl = 56, NaOMe = 48.

Acid hydrolysis of 1a yielded xylose, rhamnose and quercetin (coPC), while controlled acid hydrolysis (1.5 M HCl, 100°, 5 min) yielded quercitrin (coPC and UV spectral data).

$^{13}\text{C}$  NMR: aglycone:  $\delta$  156.84 (C-2), 135.24 (C-3), 178.60 (C-4), 161.00 (C-5), 99.20\* (C-6), 164.82 (C-7), 93.26 (C-8), 156.84 (C-9), 104.20 (C-10), 124.02 (C-1'), 116.08† (C-2'), 147.90 (C-3'), 146.82 (C-4'), 117.10† (C-5'), 120.04 (C-6'); 3-O- $\alpha$ -L-rhamnoside:  $\delta$  98.82\* (C-1), 80.40 (C-2), 70.46‡ (C-3), 71.98§ (C-4), 70.02‡ (C-5), 17.84 (Me); 4'-O- $\alpha$ -L-rhamnoside:  $\delta$  101.06 (C-1), 70.22‡ (C-2), 70.46‡ (C-3), 71.98§ (C-4), 70.02† (C-5), 18.08 (Me); 3-O- $\beta$ -D-xyloside:  $\delta$  106.40 (C-1), 73.66 (C-2), 76.02 (C-3), 70.46‡ (C-4), 65.02 (C-5). (\*, †, ‡, §: Assignments bearing the same superscript may be reversed).

$^1\text{H}$  NMR: aglycone moiety.  $\delta$  6.2 (*d*, *J* = 2.5 Hz, 6-H), 6.4 (*d*, *J* = 2.5 Hz, 8-H), 7.15 (*d*, *J* = 8 Hz, 5'-H), 7.23 (*d*, *J* = 8 Hz, 6'-H); the two lines were each broadened by 4 Hz due to coupling to 2'-H; 7.36 (*s*, *W*<sub>1/2</sub> = 4 Hz, 2'-H); sugar moieties:  $\delta$  5.27 (*s*, *W*<sub>1/2</sub> = 3.5 Hz, 1-H of rhamnoside), 5.34 (*s*, *W*<sub>1/2</sub> = 4 Hz, 1-H of rhamnoside), 2.8–4.1 (*m*, 12 sugar protons), 0.85 (*d*, *J* = 6 Hz, Me of rhamnoside), 1.08 (*d*, *J* = 6 Hz, Me of rhamnoside).

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#### REFERENCES

- Kang, S. S., Shim, S. H., Wagner, H., Mohanchari, V., Seligmann, O. and Ober-Maier, G. (1979) *Phytochemistry* **18**, 353.
- Tschesche, R., Shah, A. H. and Eckhardt, G. (1972) *Phytochemistry* **11**, 702.
- Srivastava, S. K. and Srivastava, S. D. (1979) *Phytochemistry* **18**, 1758.
- Ikram, M. and Tomlinson, H. (1976) *Planta Med.* **29**, 289.
- Wagner, H., Ertan, M. and Seligmann, O. (1974) *Phytochemistry* **13**, 857.
- Mabry, T. J., Markham, K. R. and Thomas, M. B. (1970) *The Systematic Identification of Flavonoids*. Springer, New York.
- Jurd, L. (1962) in *The Chemistry of the Flavonoid Compounds* (Geissman, T. A., ed.), p. 108. Pergamon Press, Oxford.
- Imperato, F. (1979) *Experientia* **35**, 1134.
- Harborne, J. B., Mabry, T. J. and Mabry, H. (eds.) (1975) *The Flavonoids*. Chapman & Hall, London.
- Markham, K. R., Terni, B., Stanley, R., Geiger, H. and Mabry, T. J. (1978) *Tetrahedron* **34**, 1389.
- Garcia-Granados, A. and Saenz-Burruage, J. M. (1980) *Org. Magn. Reson.* **13**, 462.
- Breitmaier, E. and Voelter, W. (1978)  $^{13}\text{C}$  NMR Spectroscopy, p. 255. Verlag Chemie, Weinheim.